crystallization papers

Acta Crystallographica Section D Biological Crystallography ISSN 0907-4449

Mohammad W. Bhuiya,^a Hideaki Tsuge,^b Haruhiko Sakuraba,^a Kazunari Yoneda,^b Nobuhiko Katunuma^b and Toshihisa Ohshima^a*

 ^aDepartment of Biological Science and Technology, Faculty of Engineering, The University of Tokushima,
 2-1 Minamijosanjimacho, Tokushima
 770-8506, Japan, and ^bInstitute for Health Sciences, Tokushima Bunri University,
 Yamashirocho, Tokushima 770-8514, Japan

Correspondence e-mail: ohshima@bio.tokushima-u.ac.jp

Crystallization and preliminary X-ray diffraction analysis of glutamate dehydrogenase from an aerobic hyperthermophilic archaeon, Aeropyrum pernix K1

Glutamate dehydrogenase from an aerobic hyperthermophilic archaeon, *Aeropyrum pernix* K1, was crystallized by the hangingdrop vapour-diffusion method using polyethylene glycol (PEG) 400 as the precipitant. The crystals belong to the hexagonal space group $P6_3$, with unit-cell parameters a = b = 98.9, c = 394.8 Å, $\alpha = \beta = 90$, $\gamma = 120^\circ$. The asymmetric unit contained one hexamer of the enzyme, giving a crystal volume per enzyme mass ($V_{\rm M}$) of 1.98 Å³ Da⁻¹ and a solvent content of 37.3%. The X-ray diffraction data were collected to a resolution of 3.0 Å at the BL6B beamline in the Photon Factory with an overall $R_{\rm sym}$ of 13.8% and a completeness of 87.1%.

1. Introduction

More than 60 strains of physiologically different hyperthermophiles have been isolated from high-temperature environments in the last two decades (Hutchins et al., 2001). Most of the hyperthermophiles growing at a temperature around the boiling point of water are known to be anaerobic organisms and obtain energy by fermentation or with a nonoxygenic respiratory system. A. pernix K1 isolated from a coastal solfataric vent is a unique hyperthermophilic archaeon that grows heterotrophically at the optimum temperature of 363-368 K under absolutely aerobic conditions (Sako et al., 1996). This is the only strictly aerobic organism growing optimally at a temperature above 363 K. Thus, the energy metabolism, synthesis and metabolism of the cell components, and properties of the enzymes of A. pernix K1 are expected to be different from many other anaerobic hyperthermophiles such as Pyrococcus furiosus and Thermococcus litoralis.

We have already investigated the structure and function of glutamate dehydrogenases (GluDHs) from marine and continental hyperthermophilic archaea (Ohshima & Nishida, 1993, 1994; Kujo & Ohshima, 1998; Kujo et al., 1999; Bhuiya et al., 2000). Recently, we purified an NADP-dependent GluDH from A. pernix K1 and cloned the gene encoding the enzyme (Bhuiya et al., 2000). We compared the primary structure of the enzyme with those of the other GluDHs from hyperthermophilic archaea and performed a phylogenetic analysis. As a result, it was revealed that the GluDH from A. pernix K1 was clustered with those from aerobic thermophiles (Sulfolobus solfataricus, S. shibatae) and the anaerobic hyperthermophile Pyrobaculum islandicum and was separated from another cluster of the

Received 21 February 2002 Accepted 8 May 2002

enzymes from Thermococcales such as Pyrococcus and Thermococcus (Bhuiya et al., 2000). The two clusters clearly reflected the difference between Crenarchaeota (the former cluster) and Euryarchaeota (the latter), the two kingdoms of Archaea. The presence of the citric acid cycle has been demonstrated in Sulfolobus (Danson, 1988), Pb. islandicum (Selig & Schönheit, 1994) and also in A. pernix K1 (Kawarabayasi et al., 1999). However, it has not yet been reported in cells of the members of the Thermococcales and the principal function of GluDHs from the Thermococcales has been suggested to be L-glutamate biosynthesis coupled with L-alanine production (Ohshima & Nishida, 1993; Kengen & Stams, 1994; Kobayashi et al., 1995). Thus, we have predicted that the physiological role of GluDHs from Sulfolobus, Pb. islandicum and A. pernix K1 is distinct from that of the Thermococcales enzymes and may be linked to the citric acid cycle via 2-oxoglutarate in the cells of these organisms (Bhuiya et al., 2000).

Recently, we demonstrated that the temperature dependence of the kinetic parameters for the GluDH from A. pernix K1 is remarkably different from that for the enzymes from Thermococcales (Bhuiya et al., 2002). The K_m values for L-glutamate of the GluDHs from P. furiosus and isolate AN1 determined at around 323 K are low compared with those determined at higher temperatures (Robb et al., 1992; Hudson et al., 1993). For A. pernix K1 GluDH, K_m decreased and the catalytic efficiency markedly increased with an increase in temperature from 323-363 K (Bhuiya et al., 2002). This suggests that the conformational state that is responsible for the kinetic properties of the enzyme is different from that of the Thermococcales enzyme. The threedimensional structures of the GluDHs from Thermococcales in Euryarchaeota have

© 2002 International Union of Crystallography Printed in Denmark – all rights reserved already been determined, i.e. GluDHs from P. furiosus (Yip et al., 1995), T. litoralis (Britton et al., 1999) and T. profundus (Nakasako et al., 2001). However, they have not been determined for the enzymes from either the species in the Crenarchaeota or the aerobic hyperthermophiles. The A. pernix K1 GluDH showed sequence identities of 51.1, 49.4 and 48.6% to GluDHs from T. profundus, T. litoralis and P. furiosus, respectively (Bhuiya et al., 2000). Comparison of the structure of the A. pernix K1 GluDH with those of the Thermococcales enzymes may help to elucidate the relationship between the physiological function and the differences in the conformational state of GluDHs from hyperthermophilic archaea. Therefore, we decided to determine the three-dimensional structure of A. pernix K1 GluDH. We now describe the crystallization and preliminary X-ray diffraction analysis of GluDH from the aerobic hyperthermophilic archaeon A. pernix K1.

2. Materials and methods

2.1. Crystallization

GluDH was purified from the crude extract of A. pernix K1 (JCM 9820) as described previously (Bhuiya et al., 2000). The purified enzyme was dialyzed against 10 mM MOPS-NaOH pH 7.0 and concentrated to 10 mg ml^{-1} for the crystallization trials. The initial screening for crystallization of the GluDH was carried out using Crystal Screens (Hampton Research, USA) by the hanging-drop vapour-diffusion method at room temperature. Crystals were grown in hanging drops consisting of 2 µl of enzyme solution with an equal volume of a reservoir solution consisting of 20%(w/v) PEG 400, 200 mM CaCl2 and 100 mM HEPES-KOH pH 7. Crystals of diffraction quality appeared within 24 h and reached maximum dimensions of $0.3 \times 0.3 \times 0.15$ mm within 2-3 d (Fig. 1).

2.2. X-ray measurements and data processing

The crystals were mounted in a capillary and diffraction data were collected using beamline BL6B ($\lambda = 1.00$ Å) at the Photon Factory (Tsukuba, Japan) using a Weissenberg camera. The crystal-to-detector distance, oscillation angle per image and coupling constant were set to 57.3 cm, 3° and 1.0, respectively. The crystals diffracted to a resolution limit of 3.0 Å. The data were processed with *DENZO* and *SCALEPACK* from the *HKL* program suite (Otwinowski

Table 1

Data collection and processing statistics of *A. pernix* K1 GluDH.

Values in parentheses are for the highest resolution shell.

| Beam source | KEK-PF BL6B |
|-----------------------------|------------------------|
| Detector | Weissenberg camera |
| Distance (cm) | 57.3 |
| Space group | $P6_3$ |
| Unit-cell parameters (Å, °) | a = b = 98.9, |
| | c = 394.8, |
| | $\alpha = \beta = 90,$ |
| | $\gamma = 120$ |
| Max. resolution (Å) | 3.0 |
| Unique reflections | 37949 |
| Redundancy | 5.2 |
| Resolution range (Å) | 50-3.0 (3.16-3.0) |
| Completeness (%) | 87.1 (87.9) |
| $R_{\rm sym}(I)$ † | 0.138 (0.297) |
| $I/\sigma(I)$ | 4.0 |

† $R_{\text{sym}}(I) = \sum_{j} \sum_{i} |I_{ij} - \langle I_{j} \rangle| / \sum_{i} \sum_{j} I_{ij}$



Figure 1

Hexagonal crystal of A. pernix K1 GluDH. The dimensions of the largest crystal are 0.3 \times 0.3 \times 0.15 mm.

& Minor, 1997). The space group was determined to be P63 or P6322 based on systematic absences. $P6_3$ was the most feasible space group judged by the statistics from SCALEPACK. A summary of the data statistics is shown in Table 1. A self-rotation function showed a sixfold axis of crystallographic symmetry; however, it did not show twofold or threefold non-crystallographic symmetry. Assuming six subunits of GluDH, each of molecular mass 46 kDa, in the asymmetric unit, the crystal volume per enzyme mass $(V_{\rm M})$ and the solvent content were calculated to be $1.98 \text{ Å}^3 \text{ Da}^{-1}$ and 37.3%, respectively. These values are within the frequently observed ranges for protein crystals (Matthews, 1968).

One hexamer occupied an asymmetric unit for GluDH from *T. profundus* (Nakasako *et al.*, 2001) and *T. litoralis* (Britton *et al.*, 1999) and one trimer for *P. furiosus* (Yip *et al.*, 1995). The crystal structure analysis is under way by the molecular-replacement method using the structures of GluDH from *T. profundus* (Nakasako *et al.*, 2001), *T. litoralis* (Britton *et al.*, 1999) and *P. furiosus* (Yip *et al.*, 1995). We have not yet obtained a clear solution, but are now attempting the use of different models and programs. Furthermore, we are also trying to solve the phases using the heavy-atom isomorphous replacement method.

We thank Drs M. Suzuki, N. Igarashi and N. Sakabe at KEK-PF for assistance with the data collection. This study was funded in part by 'The Special Scientific Research, and Pioneering Research Project in Biotechnology' promoted by the Ministry of Agriculture, Forestry and Fisheries of Japan and 'The New Energy and Industrial Technology Development Organization (NEDO) Project' promoted by the Ministry of International Trade and Industry of Japan. MWB is supported by a government scholarship from the Ministry of Education, Science and Culture of Japan.

References

- Bhuiya, M. W., Sakuraba, H., Kujo, C., Nunoura-Kominato, N., Kawarabayasi, Y., Kikuchi, H. & Ohshima, T. (2000). *Extremophiles*, 4, 333–341.
- Bhuiya, M. W., Sakuraba, H. & Ohshima, T. (2002). Biosci. Biotechnol. Biochem. 66, 873– 876.
- Britton, K. L., Yip, K. S. P., Sedelnikova, S. E., Stillman, T. J., Adams, M. W. W., Ma, K., Maeder, D. L., Robb, F. T., Tolliday, N., Vetriani, C., Rice, D. W. & Baker, P. J. (1999). *J. Mol. Biol.* 293, 1121–1132.
- Danson, M. J. (1988). Adv. Microb. Physiol. 29, 166–231.
- Hudson, R. C., Ruttersmith, L. D. & Daniel, R. M. (1993). Biochim. Biophys. Acta, 1202, 244–250.
- Hutchins, A. M., Holden, J. F. & Adams, M. W. W. (2001). J. Bacteriol. 183, 709–715.
- Kawarabayasi, Y. et al. (1999). DNA Res. 6, 83– 101.
- Kengen, S. W. M. & Stams, A. J. M. (1994). Arch. Microbiol. 161, 168–175.
- Kobayashi, T., Higuchi, S., Kimura, K., Kudo, T. & Horikoshi, K. (1995). J. Biochem. 118, 587–592.
- Kujo, C. & Ohshima, T. (1998). Appl. Environ. Microbiol. 64, 2152–2157.
- Kujo, C., Sakuraba, H., Nunoura, N. & Ohshima, T. (1999). *Biochim. Biophys. Acta*, **1434**, 365– 371.
- Matthews, B. W. (1968). J. Mol. Biol. 33, 491-497.
- Nakasako, M., Fujisawa, T., Adachi, S., Kudo, T. & Higuchi, S. (2001). *Biochemistry*, 40, 3069–3079.
 Ohshima, T. & Nishida, N. (1993). *Biosci.*
- Biotechnol. Biochem. **57**, 945–951. Ohshima, T. & Nishida, N. (1994). Biocatalysis, **11**, 117–129.
- Otwinowski, Z. & Minor, W. (1997). Methods Enzymol. 276, 307–326.
- Robb, F. T., Park, J.-B. & Adams, M. W. W. (1992). Biochim. Biophys. Acta, **1120**, 267–272.
- Sako, Y., Nomura, N., Uchida, A., Ishida, Y., Morii, H., Koga, Y., Hoaki, T. & Maruyama, T. (1996). *Int. J. Syst. Bacteriol.* 46, 1070–1077.
- Selig, M. & Schönheit, P. (1994). Arch. Microbiol.
 162, 286–294.
- Yip, K. S. P., Stillman, T. J., Britton K. L., Artymiuk, P. J., Baker, P. J., Sedelnikova S. E., Engel, P. C., Pasquo, A., Chiaraluce, R., Consalvi, V., Scandurra, R. & Rice, D. W. (1995). *Structure*, **3**, 1147–1158.